

CFD based predictive tools for liquid hydrogen hazards



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Fire and Explosion Modelling Group

We are a multi-disciplinary research group specialising in the development and validation of consequence modelling tools to address cross cutting safety issues related to energy, transport and environment.

Research areas



Gaseous hydrogen safety



Liquid hydrogen safety



Safety of lithium ion batteries



Fires in the built and natural environment



Pipeline safety



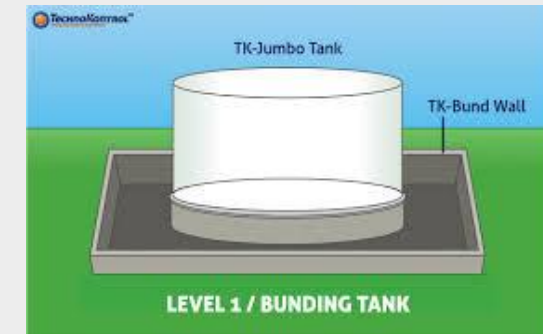
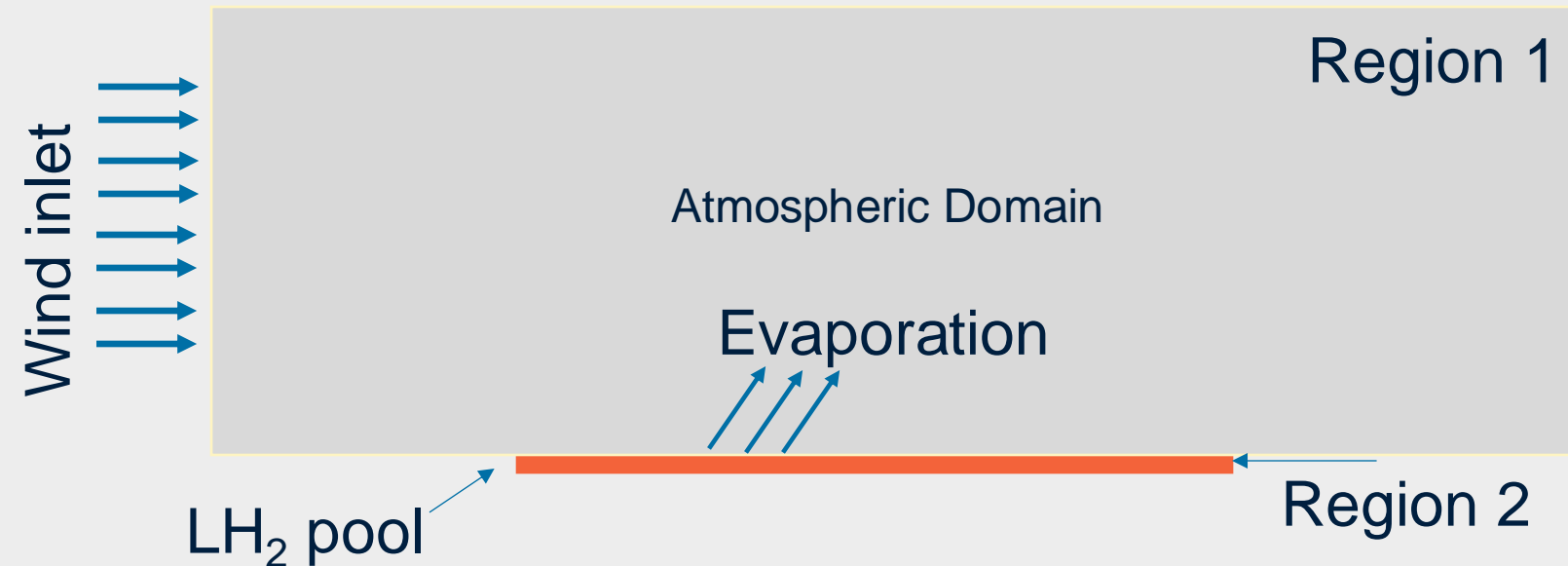
Safety of liquified natural gas

Contents

1. LH₂ vapour cloud from sudden catastrophic release
2. Unignited releases of liquid hydrogen
3. Ignited releases of hydrogen jets at cryogenic conditions
4. Vapour cloud explosions from instantaneous large-scale releases of cryogenic liquid hydrogen (LH₂)



LH₂ vapour cloud from sudden catastrophic release



<https://technokontrol.com/en/products/bunding.php>

Key assumptions:

- LH₂ flash evaporation prior to the formation of liquid pool was neglected
- Without retention pit, LH₂ spreads instantaneously to the minimum thickness of 5 cm estimated from surface roughness
- A square pool to facilitate meshing and to speed up the simulations.
- With retention pit, LH₂ content fills the pit instantaneously.
- Height of retention pit not considered.

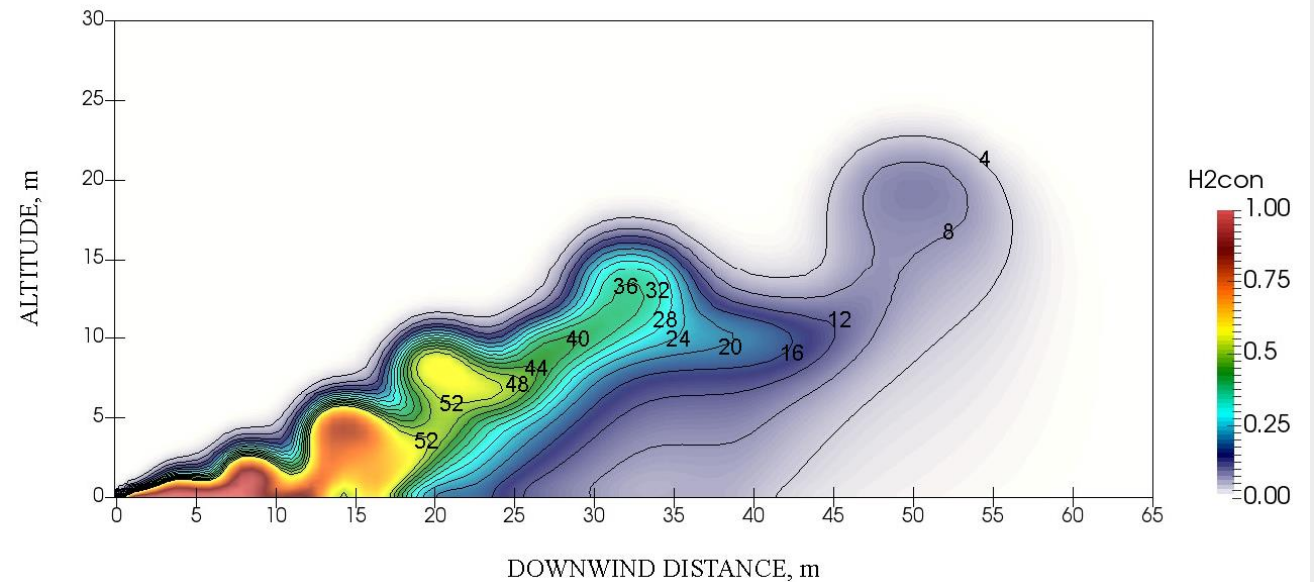
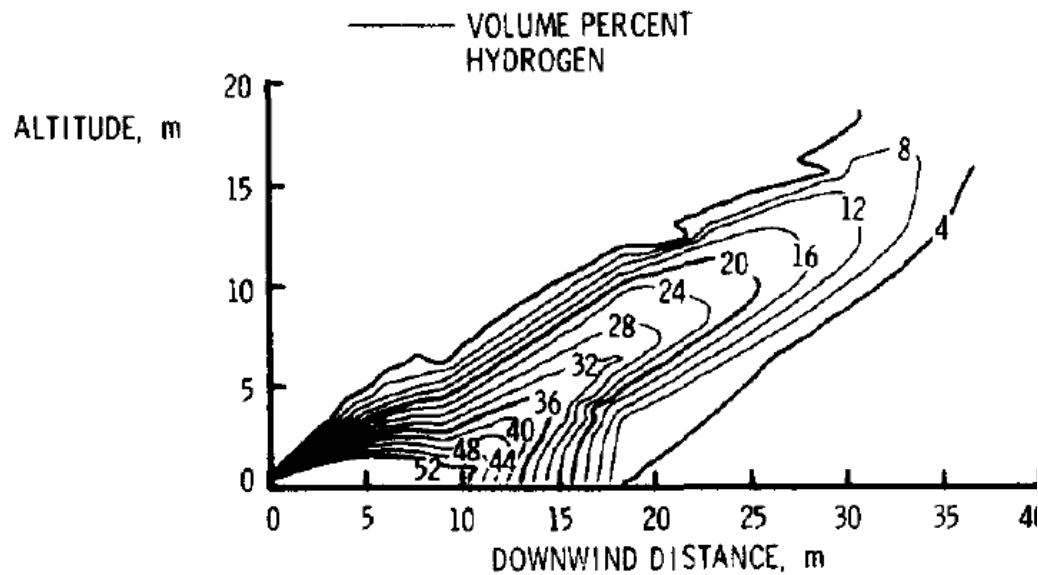
Atmospheric conditions according to Pasquill-Gifford stability:

A unstable

D neutral

F stable

Validation using NASA test 6

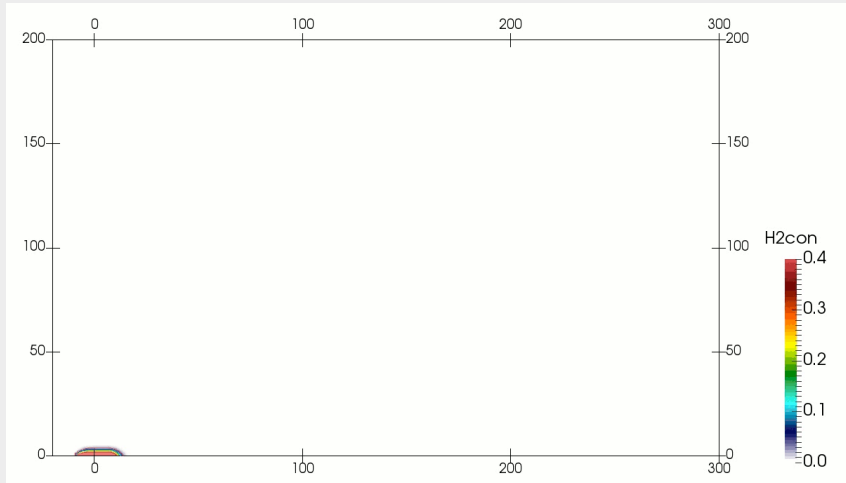


Witcofski RD, Chirivella JE. Experimental and analytical analyses of the mechanisms governing the dispersion of flammable clouds formed by liquid hydrogen spills. *Int J Hydrogen Energy* 1984; 9(5): 425-35.

Predicted H₂ molar concentration at = 20°C

	Experiment	Prediction
Horizontal extent of visible cloud	160	173
Vertical extent of visible cloud	65	69
Duration of visible cloud	90	88

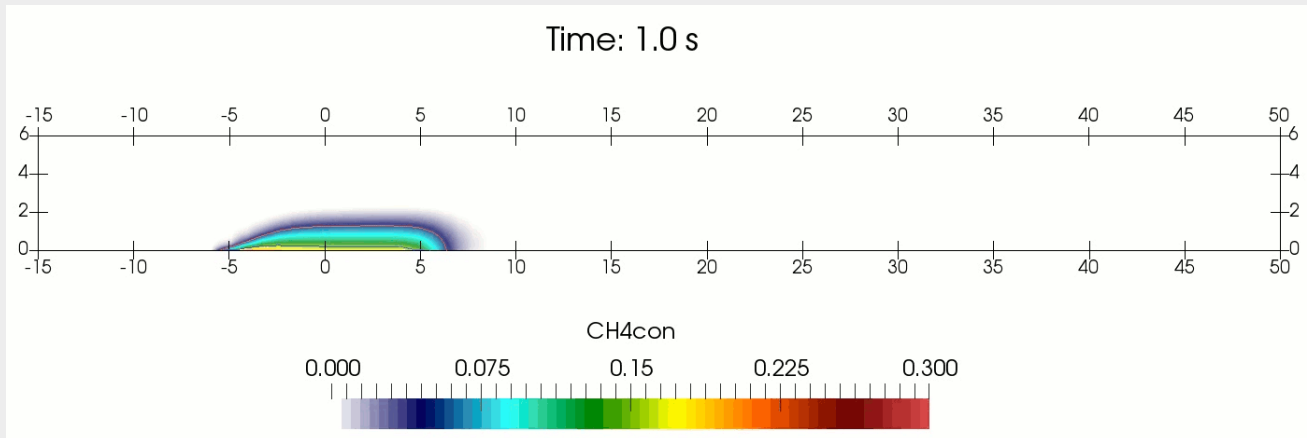
Vapour cloud from 1 ton release of LH₂ and LNG



Ambient temperature: 293K
 3 m/s stable condition

LH₂

Without retention pit (**3F**) 0 – 100
 Red line: 4% H₂ molar concentration



LNG

Without retention pit (**3F**) 0 – 150
 Red line: 5% CH₄ molar concentration.

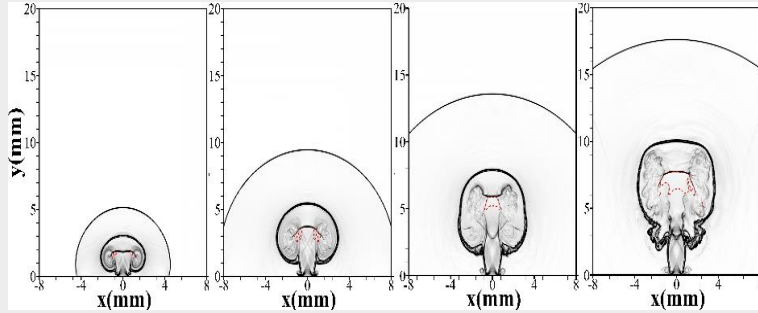
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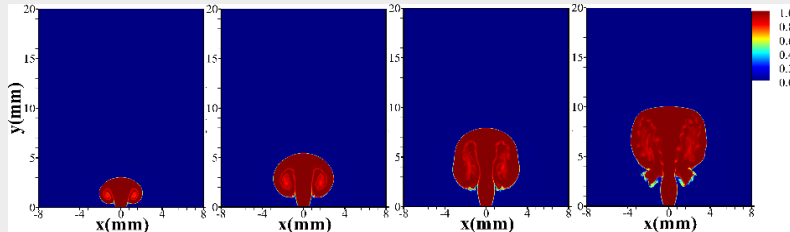
DNS of the near-field features of cryogenic jets (1)

Subtopic 1

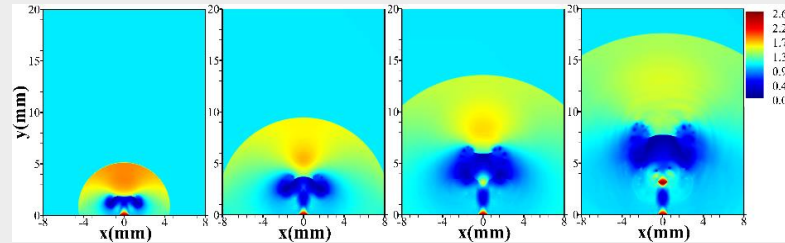
Density gradient



H2 mass fraction



Pressure

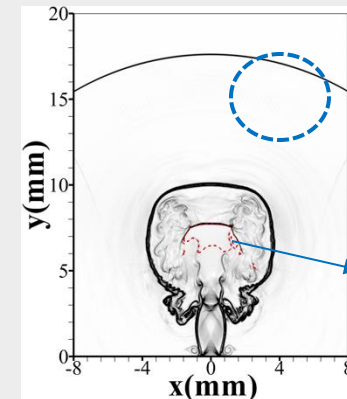
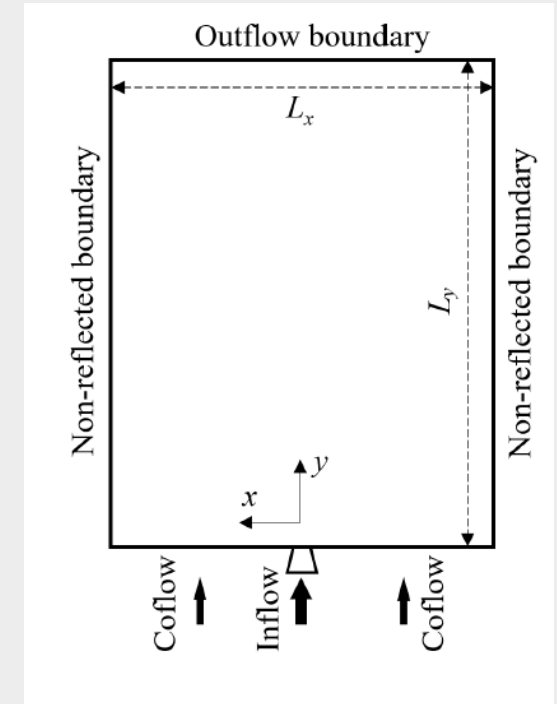


5 bar

- Transient development of the near-nozzle flow structure from $t = 10$ to $40\mu\text{s}$ shown in time interval of $10\mu\text{s}$.
- The red dashed lines denote the regions with $\text{HLP} > 0$.

The hydrogen liquefaction potentiality (HLP)

$$\text{HLP} = P_{\text{H}_2} - P_{\text{vap}}$$



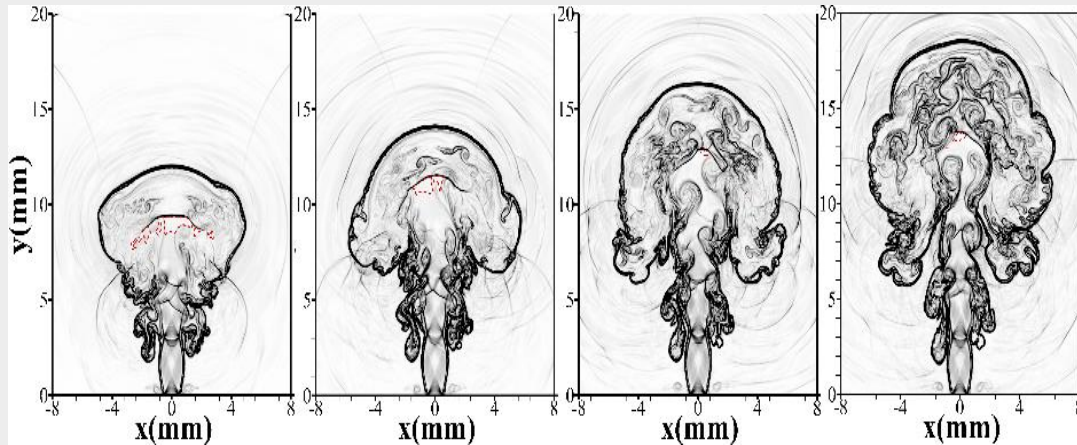
liquefaction

Ren, Zhaoxin and Wen, Jennifer X. (2020) AIP Advances, 10 (9). 095303.

DNS of the near-field features of cryogenic jets (2)

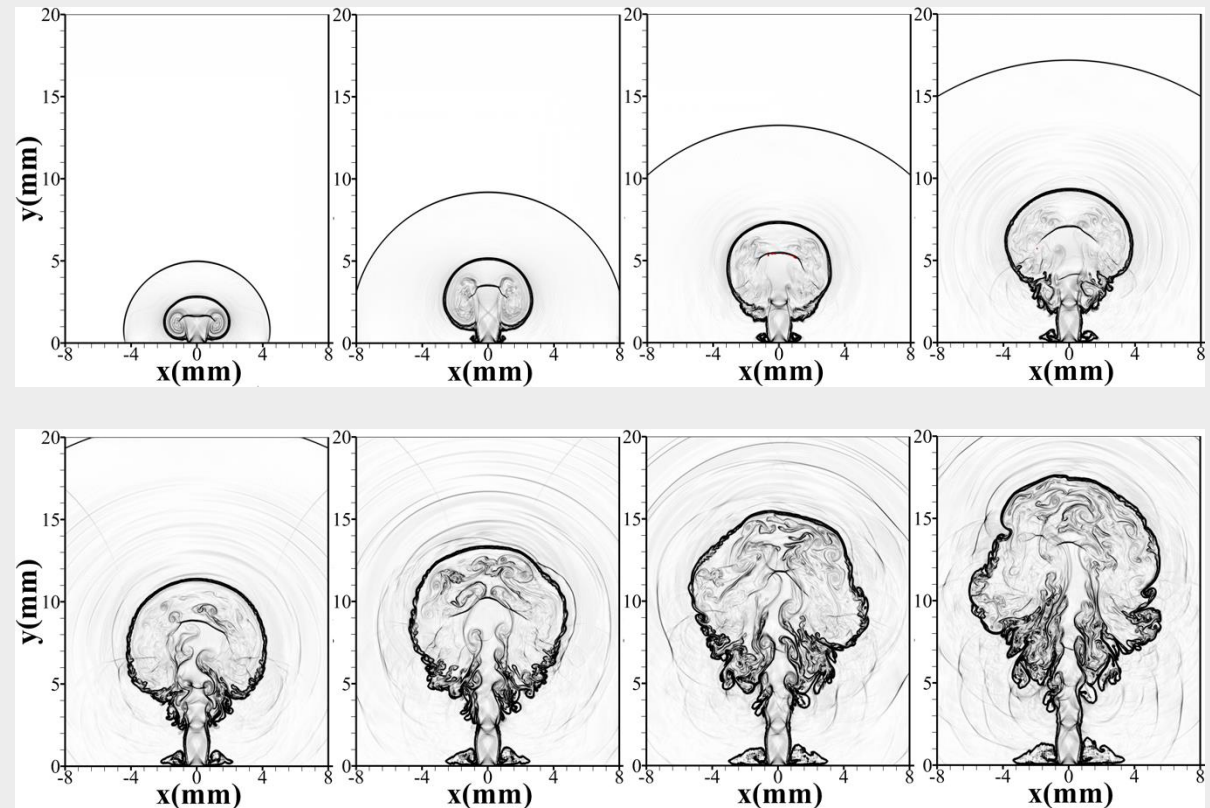
Subtopic 1

5 bar



- Instantaneous distributions of density gradient for Case HP from $t = 50$ to $80\mu\text{s}$ shown in time interval of $10\mu\text{s}$.
- The red dashed lines denote the region of $\text{HLP} > 0$.

3 bar



Low-pressure jet: no liquefaction occurs.

Contents

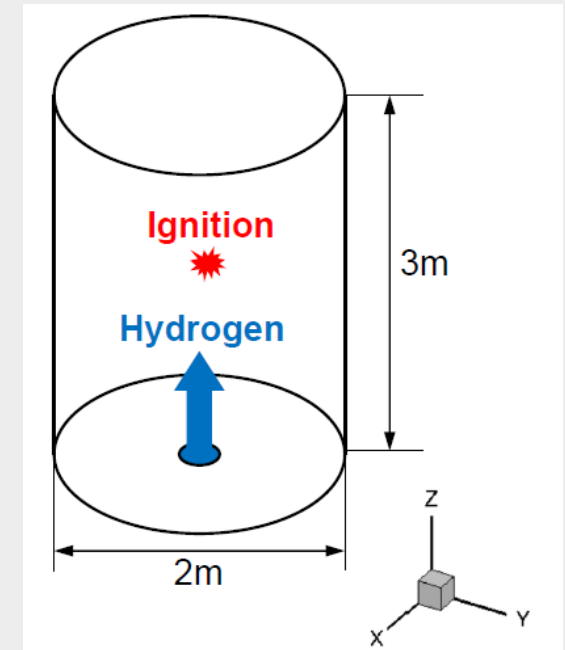
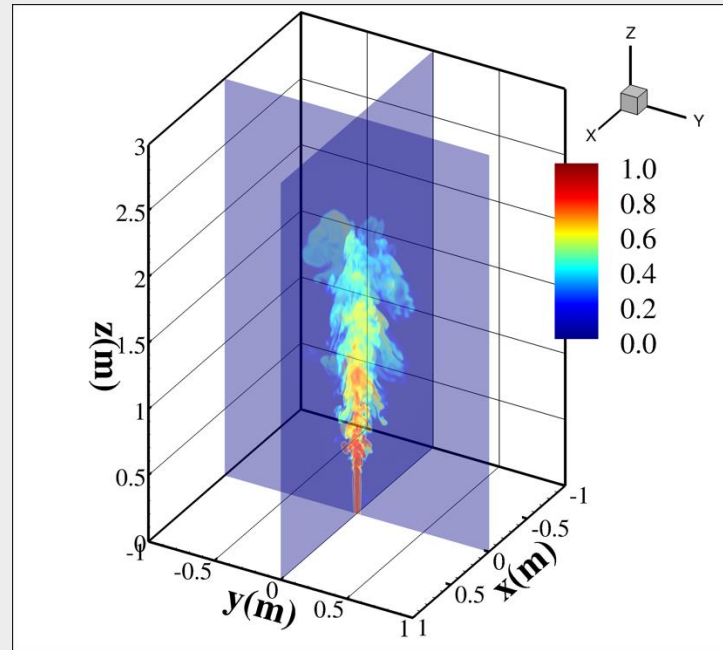
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LES of ignited releases of hydrogen jets at cryogenic conditions (1)

Subtopic 1

One equation eddy-viscosity SGS model^[1] for compressible flow
 EDC^[2] with detailed hydrogen chemistry^[3] (9 species and 19 steps) for non-premixed flame
 None-unity Lewis number effect: The molecular transport model of Burali N, et al. (2016)

Total pressure 200bar
 Total temperature 80K
 Nozzle diameter 4mm



Case	Ignition position, z (m)	Ignition temperature (K)
0.5IG	0.5	2000
1.0IG	1.0	2000
2.0IG	2.0	2000

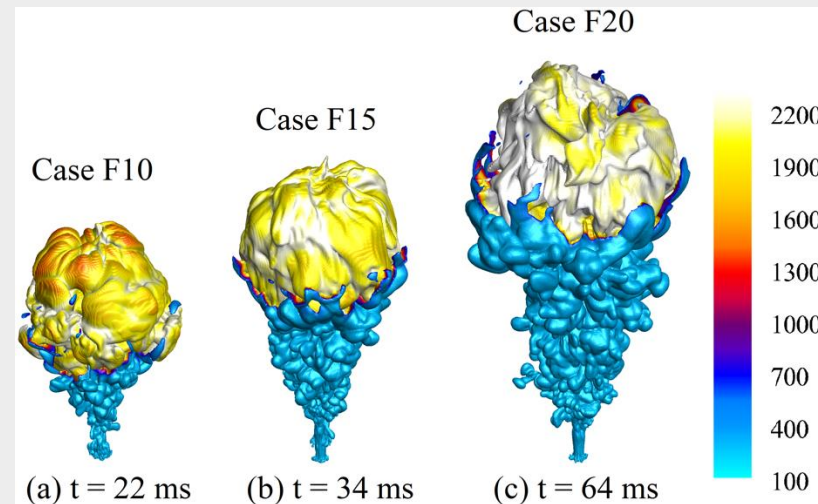
Yoshizawa A. *Physical Review E*, 1993, 48(1): 273.
 Parente A, Malik M R, Contino F, Cuoci A, Dally B B. *Fuel*, 2016, 163: 98-111.
 Ó Conaire M, Curran H J, Simmie J M, Simmie J M, Pitz W J, Westbrook C K. *Int J of Chemical Kinetics*, 2004, 36(11): 603-622.
 N. Burali, S. Lapointe, B. Bobbitt, G. Blanquart, Y. Xuan, *Combustion Theory and Modelling*, 2016, 20(4).

LES of ignited releases of hydrogen jets at cryogenic conditions (2)

Subtopic 1

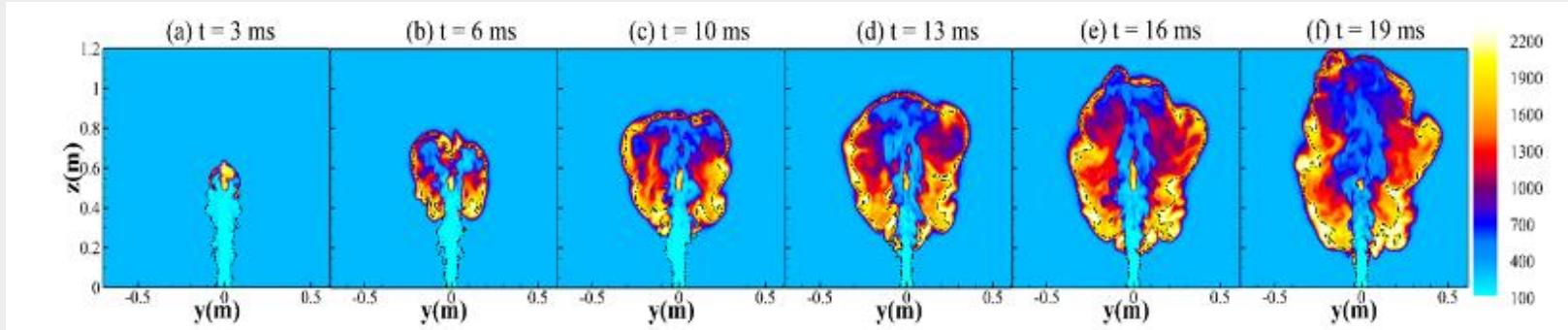
Table 1 Summary of the ignition locations considered

Case #	UF	F05	F10	F15	F20
Ignition location, z_{ig} (m)	/	0.5	1.0	1.5	2.0



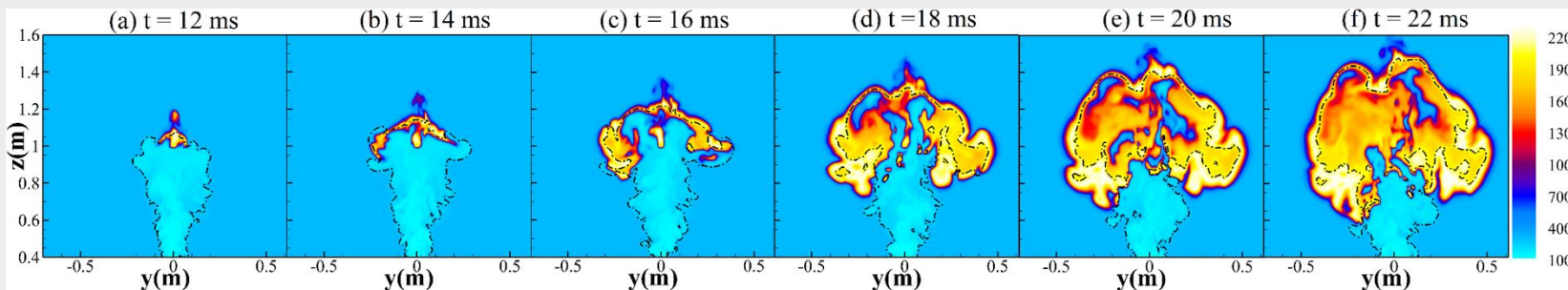
Flame structure of ignited jet for Case F10, Case F15, and Case F20 marked using $X_{H_2} = 0.04$ iso-surface colored by temperature (K).

LES of ignited releases of hydrogen jets at cryogenic conditions (4)



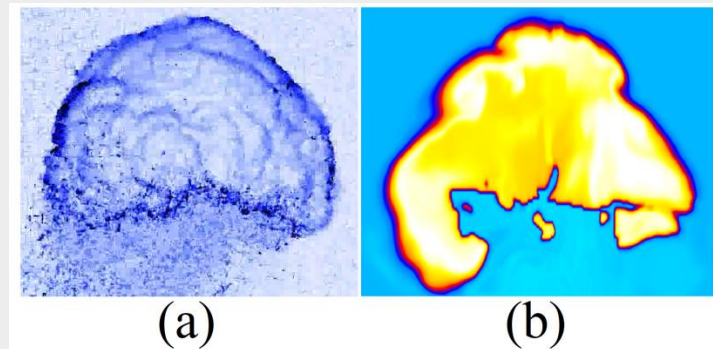
Evolution of combustion field at the y-z middle-plane for Case F05.

Here, the contours are temperature (K), and the black dashed isolines refer to hydrogen mass fraction $Y_{H_2} = 0.02$.

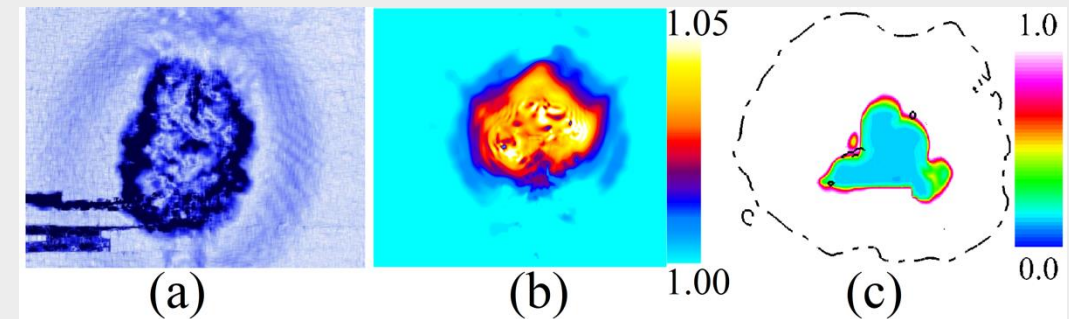


Evolution of combustion field at the y-z middle-plane for Case F10.

LES of ignited releases of hydrogen jets at cryogenic conditions (5)



Snapshots of the flame shapes: (a) experiment (reproduced from Friedrich et al. (2021)) and (b) Case F20.



Snapshot of the deflagration waves: (a) experimentally observed (reproduced from [20]), (b) predicted pressure contour for Case F15, (c) predicted density contour (kg/m^3) with static pressure iso-line of 1.03 atm.

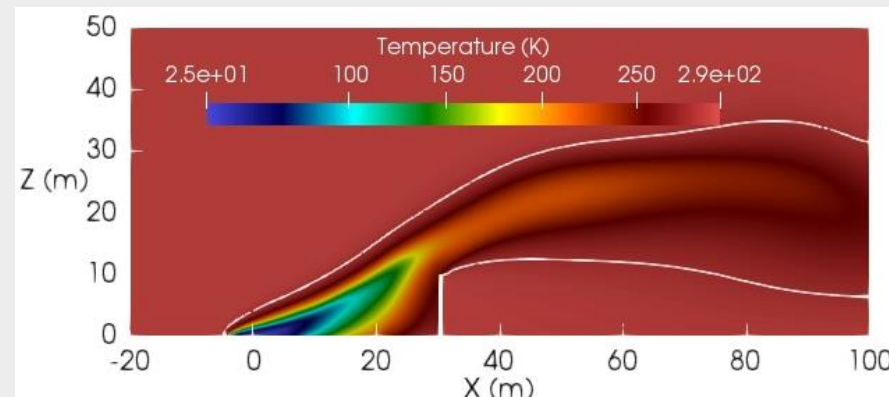
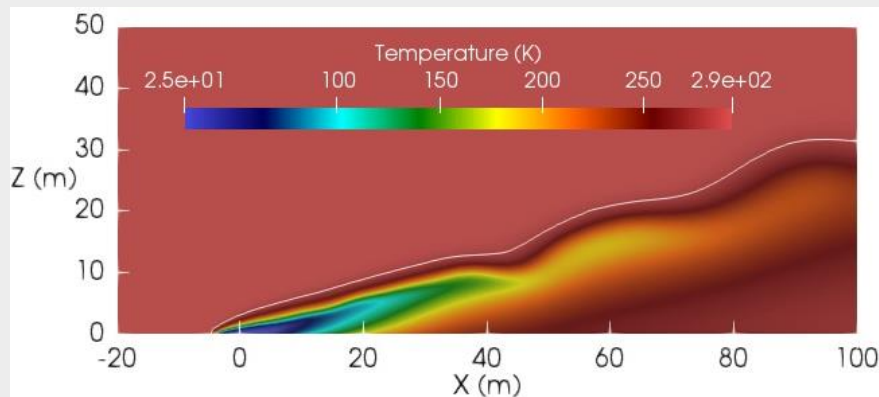
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Vapour cloud explosions from instantaneous large-scale releases of cryogenic liquid hydrogen (1)

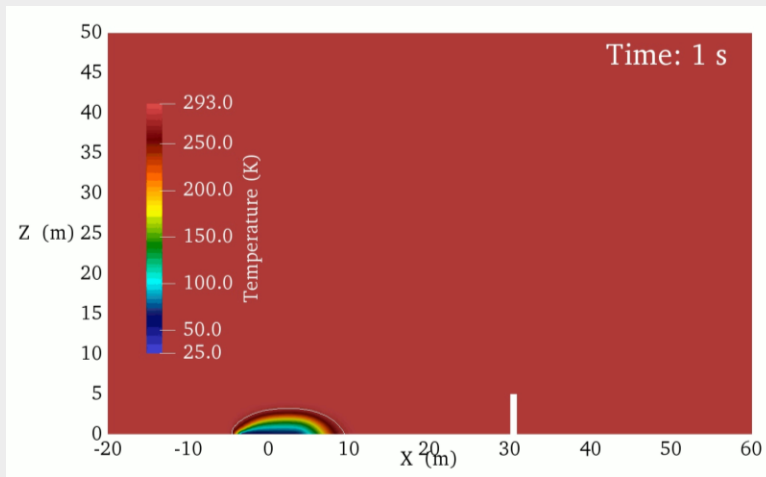
Pool size	$5 \times 5 \text{ m}^2$
Mass	600 kg LH ₂
Wind	3 m/s
Temperature	20.4 K
Barrier walls	5 or 10 m at 30 or 40 m from pool centre.
Ignition	40 s for a duration of 0.5 s.

Predicted temperature for the cases with and without a barrier just prior to the ignition



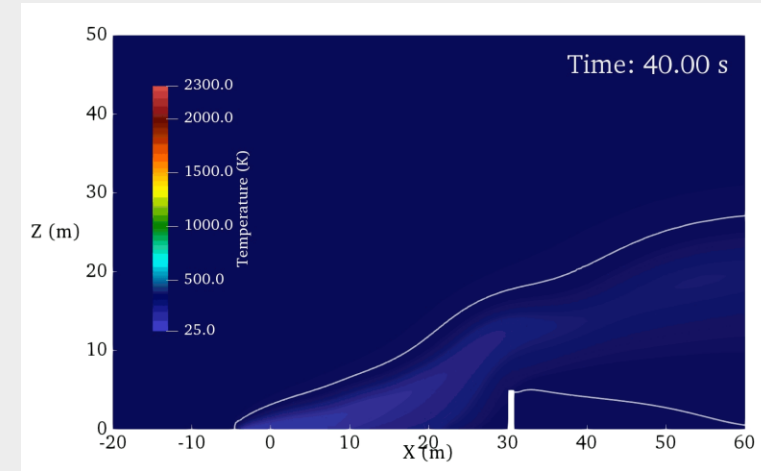
Vapour cloud explosions from instantaneous large-scale releases of cryogenic liquid hydrogen (2)

Cloud before ignition

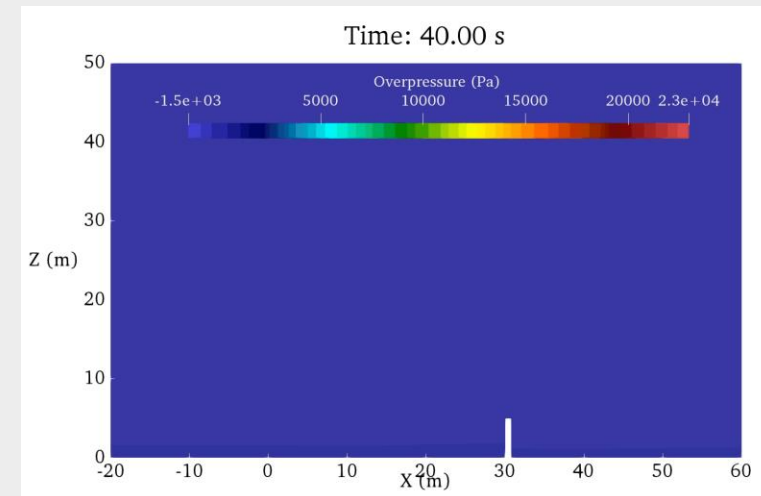


Post ignition

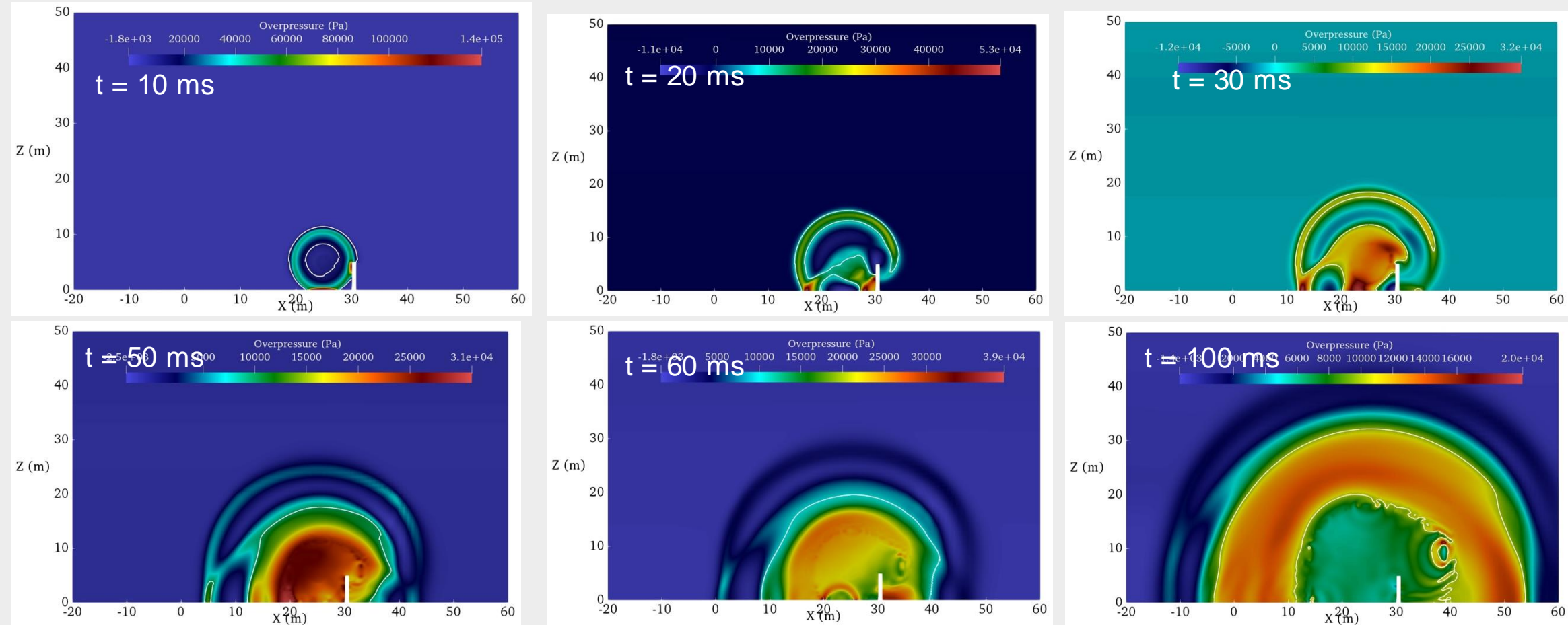
Temperature



Overpressure



Vapour cloud explosions from instantaneous large-scale releases of cryogenic liquid hydrogen (3)



Pressure contours in the middle plane at 10 ms, 20 ms, 30 ms, 50 ms, 60 ms, 100 ms after the ignition.

Concluding remarks

- The use of hydrogen as aviation fuels brings new challenges associated with accidental releases and ignition.
- Further knowledge gaps also exist the hydrogen is most likely to be stored onboard in its liquid form.
- If potential releases (*united/ignited*) exit the aircraft, knowledge gaps also exist about how they affect the aerodynamics of the aircraft and its contrails.